

2-[(2-Chlorophenyl)(1,2,3,4-tetrahydro-1-oxo-2-naphthyl)methyl]malononitrile

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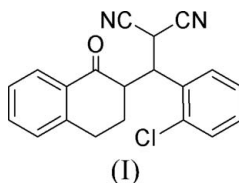
Key indicators

Single-crystal X-ray study
T = 193 K
Mean $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$
R factor = 0.064
wR factor = 0.165
Data-to-parameter ratio = 17.7For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound, $\text{C}_{20}\text{H}_{15}\text{ClN}_2\text{O}_3$, was synthesized by the reaction of 2-(2-chlorobenzylidene)-1-tetralone and malononitrile in dimethylformamide in the presence of KF–montmorillonite as a catalyst. Intermolecular $\text{C}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{N}$ hydrogen bonds link adjacent molecules into dimers and propagate chains along the *c* axis.

Comment

Recently, montmorillonite, in both natural and exchanged forms, has received considerable attention as a catalyst in various organic syntheses because of its environmental compatibility, low cost, high selectivity, reusability and operational simplicity (Hauke *et al.*, 1994). Typical reactions include rearrangements (Chalais *et al.*, 1986), oxidation (Cornelis & Laszlo, 1980) and addition (Braibante *et al.*, 1994). In most previous work, montmorillonite clays have been reported as being used as acid catalysts (Villemin & Martin, 1994; Cornelis *et al.*, 1984). However, we have found that montmorillonite coated with potassium fluoride (KF–montmorillonite), as an alkaline catalyst, results in higher selectivity, lower cost and easier work-up in organic synthesis (Shi *et al.*, 2002). The title compound, (I), was synthesized by the reaction of 2-(2-chlorobenzylidene)-1-tetralone and malononitrile in dimethylformamide in the presence of KF–montmorillonite and its structure is reported here (Fig. 1).



In the molecule of (I), the fused six-membered C2–C7 ring adopts a half-chair conformation. Atoms C3, C4, C5 and C6 are coplanar, with a maximum deviation of 0.010 (2) Å for atom C5, while atoms C2 and C7 deviate from this plane by 0.130 (2) and -0.564 (2) Å, respectively. Similar conformations were observed in 2-amino-4-phenyl-5,6-dihydrobenzo[*h*]quinazoline (Wang, Shi *et al.*, 2003) and 2-amino-3-cyano-4-(4-methoxyphenyl)-1,4,5,6-tetrahydrobenzo[*h*]chromene (Wang, Li *et al.*, 2003). The plane through C3/C4/C5/C6 is almost parallel to that of the fused benzene ring C4/C5/C8–C11, with a dihedral angle of 1.53 (2)°.

In the crystal structure of (I), intermolecular $\text{C}20-\text{H}20\cdots\text{O}1^i$ and $\text{C}12-\text{H}12\cdots\text{O}1^i$ hydrogen bonds link adjacent molecules into inversion-related dimers. Additional $\text{C}10-\text{H}10\cdots\text{N}1^{ii}$ interactions form a network structure (Fig. 2; symmetry codes as in Table 1).

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Experimental

The title compound, (I), was prepared by the reaction of 2-(2-chlorobenzylidene)-1-tetralone (0.54 g, 2 mmol) and malononitrile (0.13 g, 2 mmol) in the presence of KF–montmorillonite (0.25 g) in dimethylformamide (10 ml) at room temperature for 9 h (yield 82%; m.p. 453–455 K). Crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of a solution in dimethylformamide. Elemental analysis, calculated: C 71.75, H 4.52, N 8.37%; found: C 71.84, H 4.47, N 8.52%. IR (KBr, ν , cm^{-1}): 3065 (Ar–H), 2950, 2921, 2868 (C–H), 2247 (CN), 1673 (C=O), 1597, 1476, 1435 (phenyl ring).

Crystal data

$\text{C}_{20}\text{H}_{15}\text{ClN}_2\text{O}$	$V = 1695.2 (3) \text{ \AA}^3$
$M_r = 334.79$	$Z = 4$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 8.2263 (8) \text{ \AA}$	$\mu = 0.23 \text{ mm}^{-1}$
$b = 19.3791 (19) \text{ \AA}$	$T = 193 (2) \text{ K}$
$c = 10.6624 (10) \text{ \AA}$	$0.80 \times 0.60 \times 0.19 \text{ mm}$
$\beta = 94.198 (3)^\circ$	

Data collection

Rigaku Mercury CCD area-detector diffractometer	18644 measured reflections
Absorption correction: multi-scan (Jacobson, 1998)	3859 independent reflections
$T_{\min} = 0.835$, $T_{\max} = 0.957$	3452 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.024$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.064$	218 parameters
$wR(F^2) = 0.165$	H-atom parameters constrained
$S = 1.10$	$\Delta\rho_{\text{max}} = 0.43 \text{ e \AA}^{-3}$
3859 reflections	$\Delta\rho_{\text{min}} = -0.56 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D\text{---}H\cdots A$	$D\text{---}H$	$H\cdots A$	$D\cdots A$	$D\text{---}H\cdots A$
$\text{C}20\text{---}H20\cdots O1^i$	0.95	2.52	3.277 (3)	136
$\text{C}20\text{---}H20\cdots O1$	0.95	2.46	3.227 (4)	138
$\text{C}12\text{---}H12\cdots O1^i$	1.00	2.50	3.489 (3)	170
$\text{C}10\text{---}H10\cdots N1^{ii}$	0.95	2.62	3.417 (3)	141

Symmetry codes: (i) $-x, -y + 1, -z + 1$; (ii) $-x, -y + 1, -z$.

H atoms were positioned geometrically and refined as riding, with $\text{C---H} = 0.95 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for aromatic H, $\text{C---H} = 1.00 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for CH, and $\text{C---H} = 0.99 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for CH_2 .

Data collection: *CrystalClear* (Rigaku, 1999); cell refinement: *CrystalClear*; data reduction: *CrystalStructure* (Rigaku/MS, 2003); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997a); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997a); molecular graphics: *SHELXTL* (Sheldrick, 1997b); software used to prepare material for publication: *SHELXL97*.

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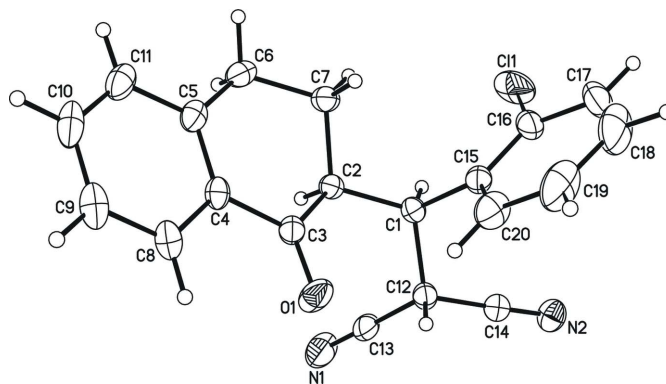


Figure 1

The molecular structure of (I), showing 30% probability displacement ellipsoids and the atom-numbering scheme.

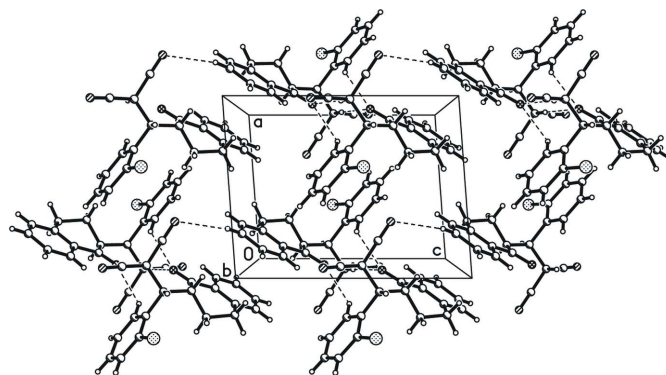


Figure 2

A packing diagram for (I), with hydrogen bonds drawn as dashed lines.

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